

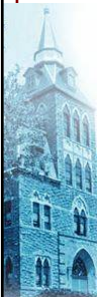
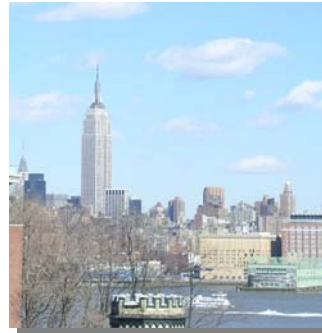
A Look At Microchemical Systems

What are MCS? What are some key limitations?

Prof. R.S. Besser

Chemical Engineering
Stevens Institute of Technology, Hoboken, NJ

Stevens Institute of Technology



Founded 1871

Stevens family: Commercialized urban ferry transport in NYC

4500 students: 1700 trad. undergrad; 2800 grad (most p-t)

Freshmen: 3.8 GPA; SAT: 1200-1400 (25%-75%)

Chem/Biomed/Matls Engineering → *NJ Center for Microchemical Systems*

New Jersey Center for MicroChemical Systems (NJCMCS)



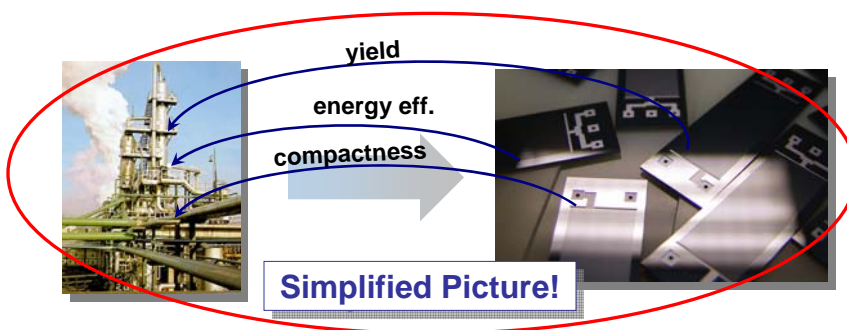
- Official start in September 2002
 - \$10.0M commitments to date
- Vision
 - Leadership for microchemical device/fundamental understanding, design methodology and tools development
- Mission
 - Original research, education of new PhDs
- Systems-level concept demo with partners
 - Army-Picatinny, Bristol-Myers Squibb, FMC, Lucent-Bell Labs/NJNC
 - Applications: portable power, pharmaceutical processing, critical industrial chemicals, biomedicine



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Microchemical Systems

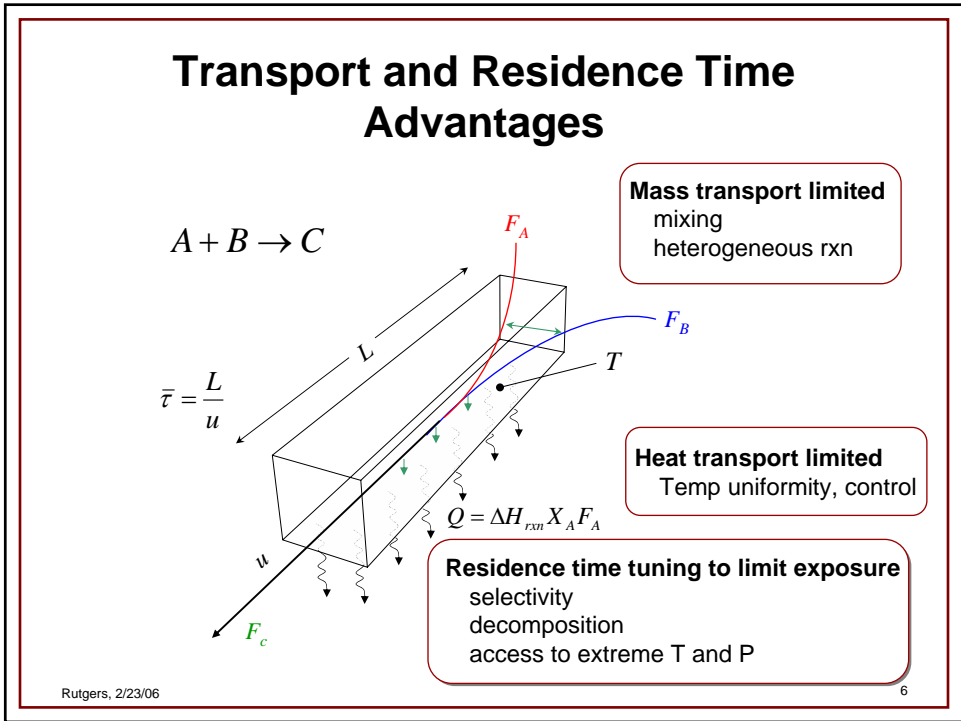
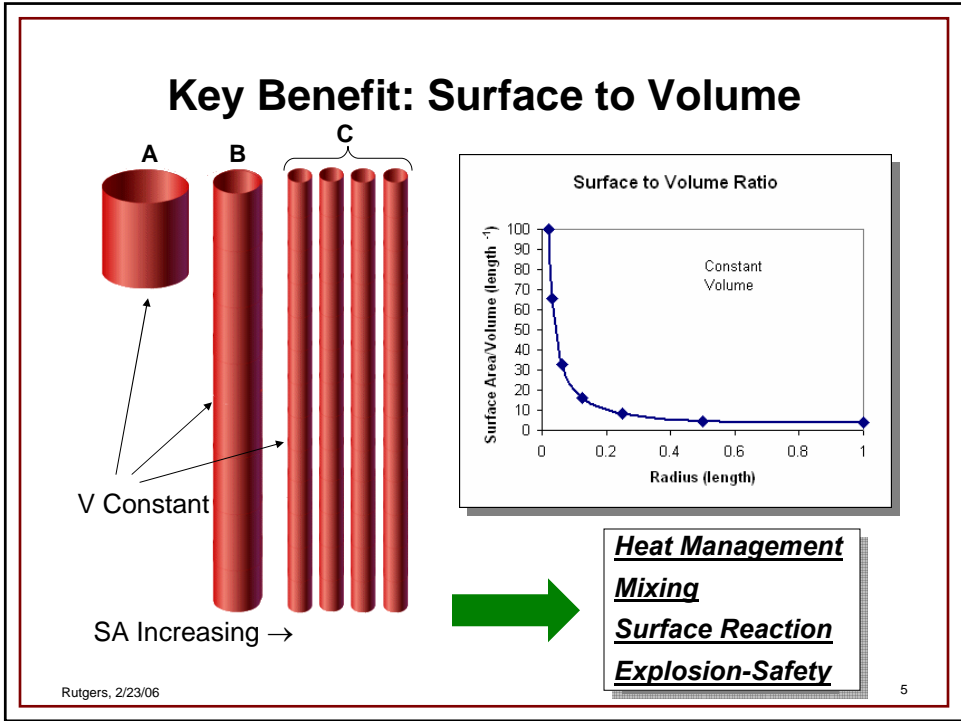


Miniature reaction and other unit operations, possessing *specific advantages* over conventional chemical systems

Not Lab-on-Chip: chemical production vs. analysis

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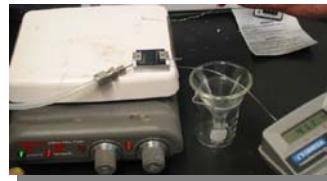
Surface to Volume: Superb Transport

Example: Overall Heat Transfer Coefficient

| <i>Hx Type</i> | <i>U (W/m²K)</i> |
|----------------|-----------------------------|
| Tubular | 150-1200 |
| Spiral | 700-2500 |
| Plate | 1000-4000 |

Microchannel: 3800-6800 W/(m²K)

(Stevens undergrad design project)



(500x500 μm^2 x 1.5 cm channels)

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- SIT and NJCMCS
- MCS: a Definition

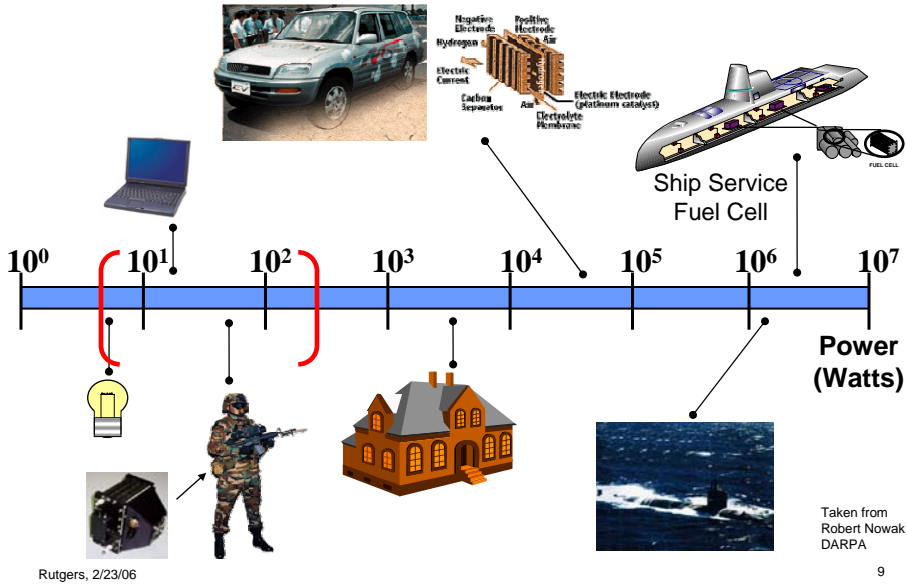
Key Limitations in MCS and Examples of How We Are Exploring Them

- **Geometry Limitation in Heat Removal**
- Thermal Integration Limitation in a Multiunit System
- Multiphase Mixing Limitation
- Integrated Process Control Limitation

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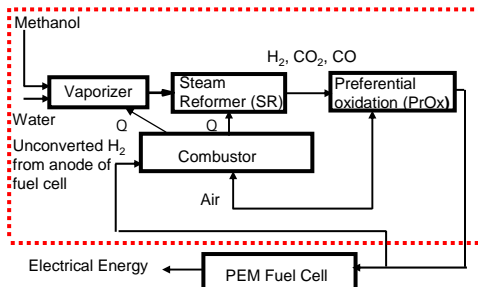
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Fuel Cells: Applications & Power Ranges



Next Generation Portable Power Sources Fuel Cells & Fuel Processing

- ❖ As alternatives to batteries
- ❖ Offers high energy density. Allows portable devices to operate for longer times with less recharging



Methanol to hydrogen: Components of fuel processor

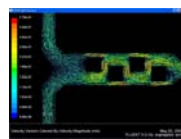
- ❖ Generating hydrogen by processing of easily stored liquid hydrocarbons
- ❖ Methanol reforming: attractive source of H₂ for portable PEMFC

Geometry Limitation in Heat Removal

- Exothermic reactions at high processing rates generate lots of heat
- Thermal resistance leads to temperature non-uniformities (“hot spots”)
- Critical reactions are sensitive to temperature variations
- Can MCS help?--How small is small enough?

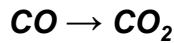
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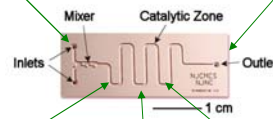
CFD Simulation

Looking for Improvements with MCS: Preferential Oxidation of CO₂



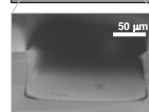
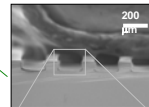
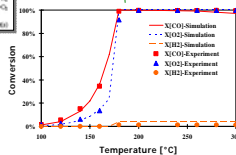
Microkinetic Array

Si Test Reactor



Chemical Kinetics Simulation

| | |
|---|---|
| 1. H ₂ O(g) ⇌ H ₂ (g) + 1/2 O ₂ (g) | 15. O ₂ (g) ⇌ 2 O(g) |
| 2. CO(g) ⇌ C(s) + O(g) | 16. H ₂ O(g) ⇌ H ₂ (g) + O(g) |
| 3. SiO ₂ (s) ⇌ Si(s) + O ₂ (g) | 17. CO(g) ⇌ C(s) + O(g) |
| 4. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 18. H ₂ O(g) ⇌ H ₂ (g) + O(g) |
| 5. CO(g) ⇌ C(s) + O(g) | 19. CO(g) ⇌ C(s) + O(g) |
| 6. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 20. H ₂ (g) ⇌ 2 H(g) |
| 7. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 21. CO(g) ⇌ C(s) + O(g) |
| 8. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 22. O ₂ (g) ⇌ 2 O(g) |
| 9. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 23. O ₂ (g) ⇌ 2 O(g) |
| 10. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 24. O ₂ (g) ⇌ 2 O(g) |
| 11. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 25. H ₂ O(g) ⇌ H ₂ (g) + O(g) |
| 12. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 26. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) |
| 13. CO ₂ (g) ⇌ CO(g) + 1/2 O ₂ (g) | 27. H ₂ (g) ⇌ H ₂ (g) |
| 14. H ₂ O(g) ⇌ H ₂ (g) + 1/2 O ₂ (g) | 28. H ₂ O(g) ⇌ H ₂ (g) + 1/2 O ₂ (g) |

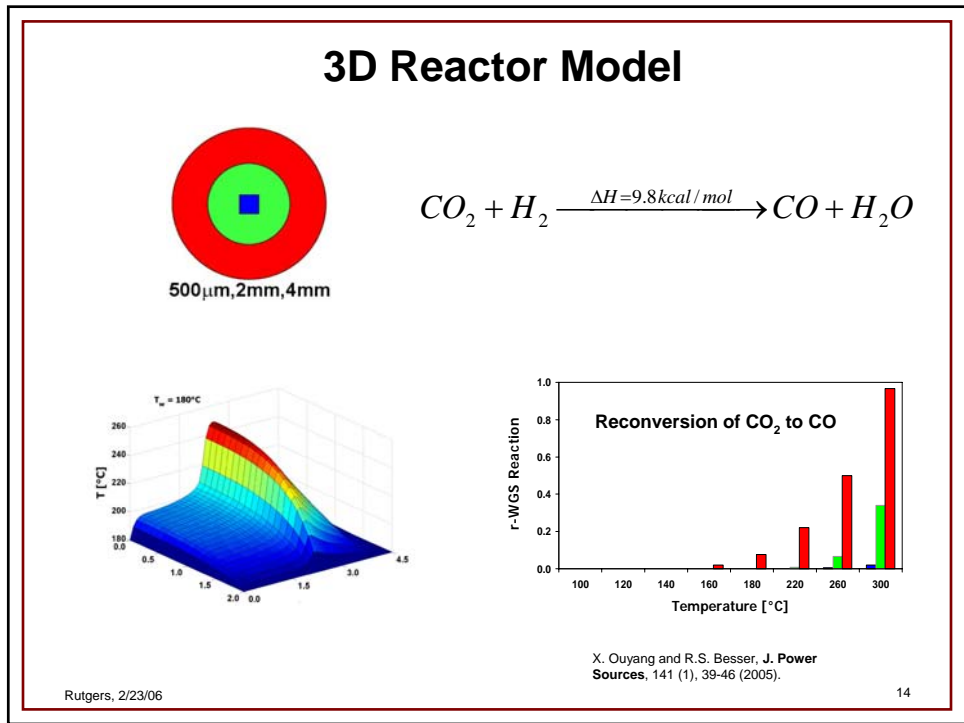
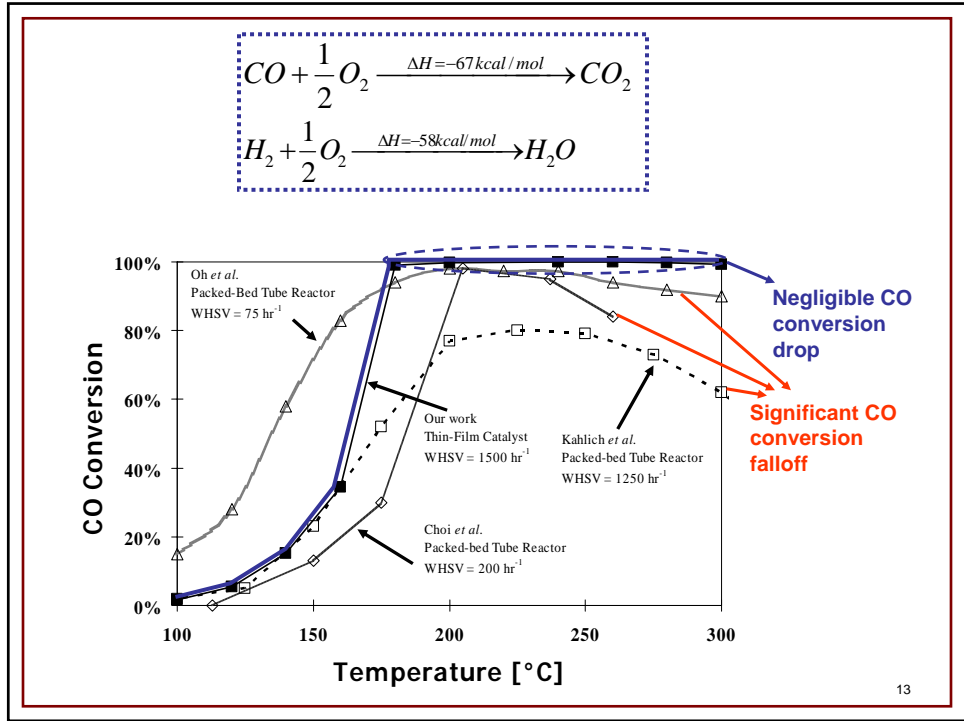


Thin-Film Catalyst Coating

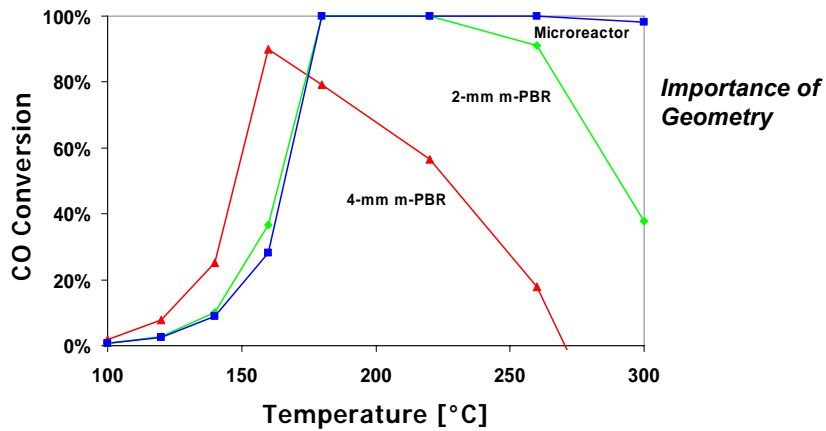
X. Ouyang, L. Bednarova, P. Ho, and R.S. Besser, *AIChE J.*, 51 (6), 2005.

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Temperature Dependence of X_{CO}



Greater diameter



More severe hot spot formation



Greater r-WGS activity

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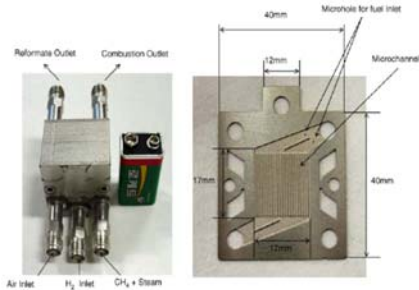
Key Limitations in MCS and Examples of How We Are Exploring Them

- Geometry Limitation in Heat Removal
- Thermal Integration Limitation in a Multiunit System
- Multiphase Mixing Limitation
- Integrated Process Control Limitation

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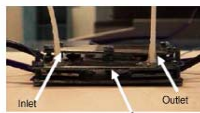
Microfabricated reactors are flat and sheet-like



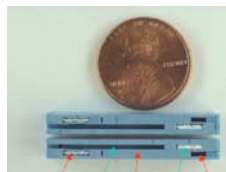
Assembled cat. Combustor/reformer (S.K. Ryi et al. Chemical Engineering Journal 113 2005 47-53)



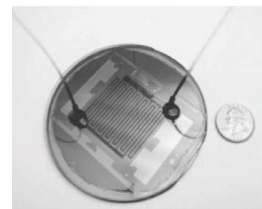
Fuel processor module. L. Thompson, Univ. Mich.



Vaporizer-reformer stack (O.J. Kwon et al.)



Solid State Research Center, Motorola Labs

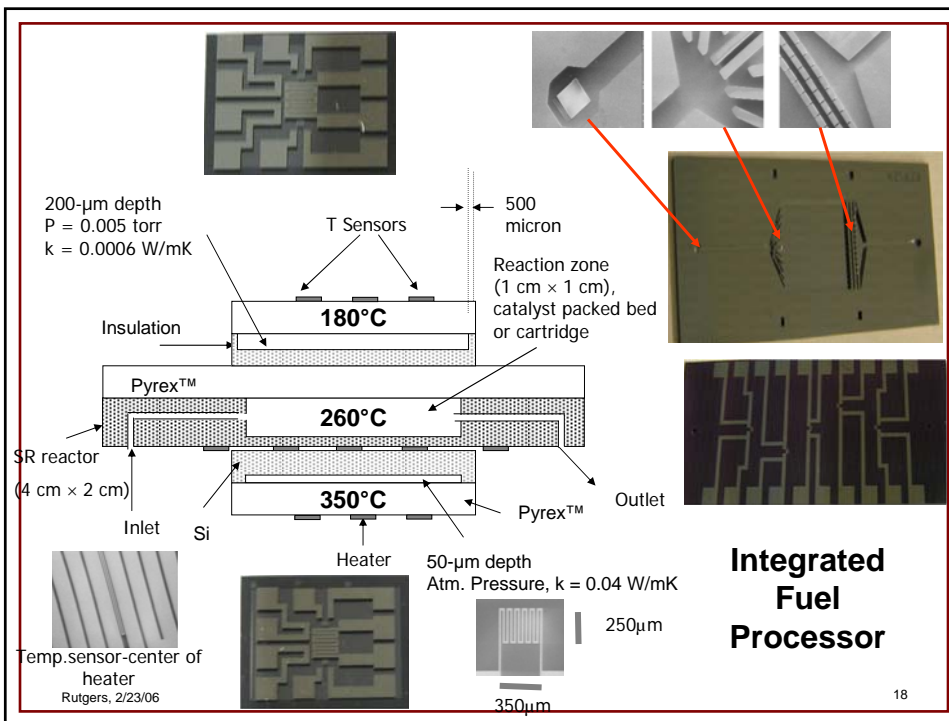


Pattekar-Kothare, IEEE-JMEMS

1. High Surface to Volume
2. Lossy, bad for heat retention
3. Good for inter-unit heat transfer
4. But units need to operate at their individual optimal temperatures

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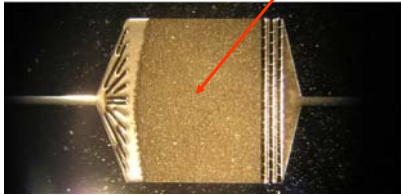
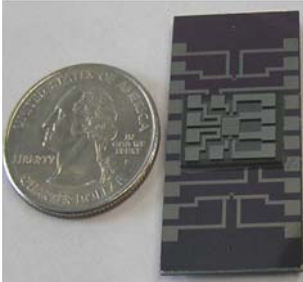

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Packed bed of catalyst

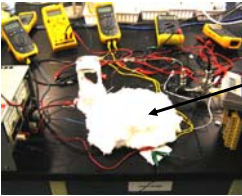




Incorporation of catalyst in the form of packed bed by vacuum loading. Catalyst loading achieved: 51 mg


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Thermal Characterization

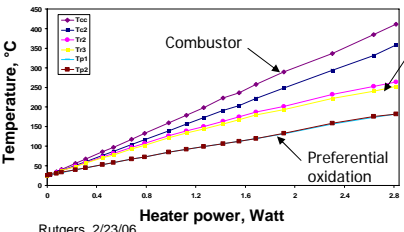
Thermal Characterization: Measure critical thermal parameters like Q required, transfer of heat between components, temperature profiles in different components, insulation effectiveness, and heat loss mechanisms



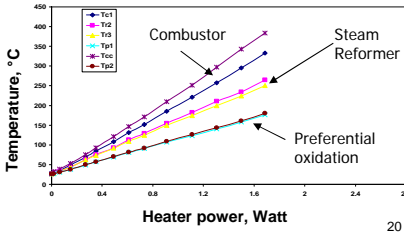
Ambient pressure testing. Device placed in bowl of insulation



Vacuum testing to distinguish multiple modes of heat transport in the system by eliminating convective losses

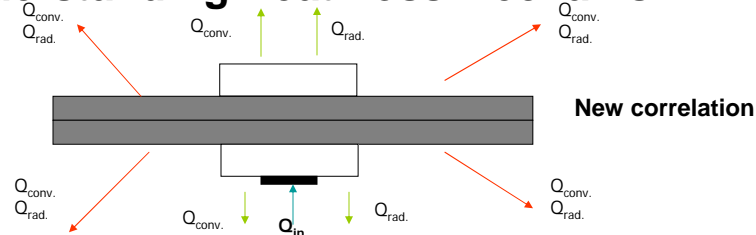


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Thermal Characterization Understanding Heat Loss Mechanism



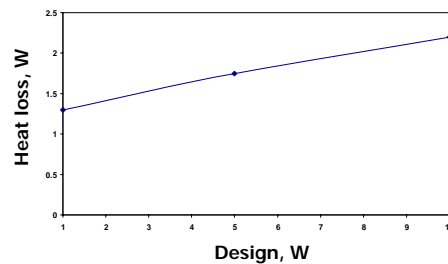
| Experiments | Q_{total} | T_{comb} | $T_{reactor}$ | T_{PrOx} | $Q_{cond.}$ | $Q_{conv.}$ | $Q_{rad.}$ | $h_{conv.}$ |
|--|-------------|------------|---------------|------------|-------------|-------------|------------|-------------|
| Vacuum suspended | 3.4 | 443 | 260 | 181 | — | 0.3 | 2.9 | 1.5 |
| Atm. suspended | 5.6 | 517 | 259 | 166 | — | 2.6 | 3.1 | 10.3 |
| Vacuum with Al foil | 2.2 | 404 | 262 | 190 | — | 0.3 | 1.9 | 1.5 |
| Atm. with Al foil | 4.7 | 476 | 261 | 166 | — | 1.9 | 2.8 | 10.3 |
| Vacuum, covered with fiberglass insulation | 1.7 | 386 | 263 | 181 | 1.5 | 0.2 | — | 0.5 |
| Atm., covered with fiberglass insulation | 2.8 | 411 | 263 | 181 | 1.7 | 1.1 | — | 7.0 |

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Thermal Characterization: Conclusion

1. Both radiation and convection important
2. Combination insulation (vac, low-k, reflective) is most effective (insulation design)
3. Think about 3D architectures instead of sheets (packaging)



| Design (W) | Loss (W) |
|------------|----------|
| 1.3 | 1.3 |
| 5 | 1.8 |
| 10 | 2.2 |

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Key Limitations in MCS and Examples of How We Are Exploring Them

- Geometry Limitation in Heat Removal
- Thermal Integration Limitation in a Multiunit System
- **Multiphase Mixing Limitation**
- Integrated Process Control Limitation

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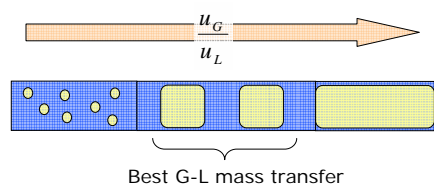
23

Multiphase (S,L&G) Mixing Is Difficult Without Turbulence

And the pressure drop becomes enormous

Two-phase flow

Bubble ↔ Taylor/Slug ↔ Annular

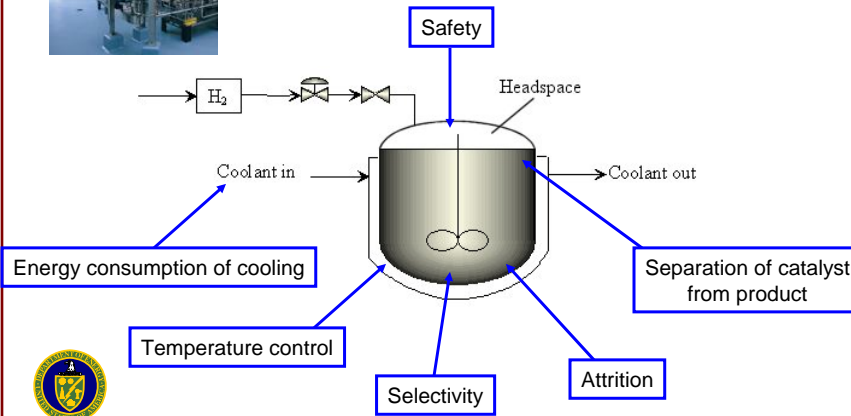
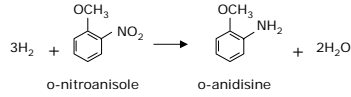


- What flow regimes exist within the microreactor?
- How do flow regimes affect the reaction conversion?
- Can we design a microreactor to operate in bubble mode and still be kinetically-controlled?

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Pharmaceutical Hydrogenation

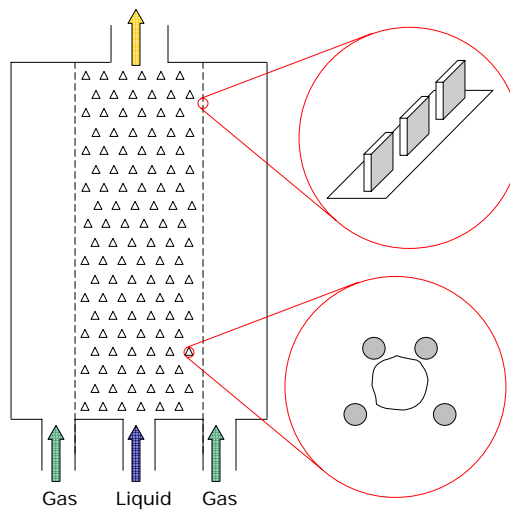


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Reactor Concept

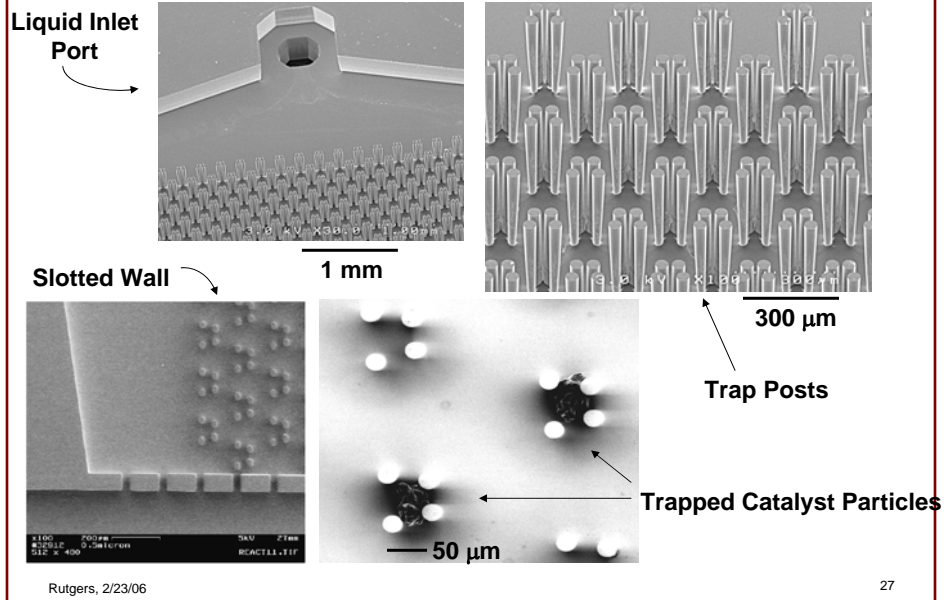
- Separate gas & liquid channels
- Slotted walls-allow gas access along entire length
- Catalyst traps-engineering porosity for ΔP
- Access to range of u_G/u_L



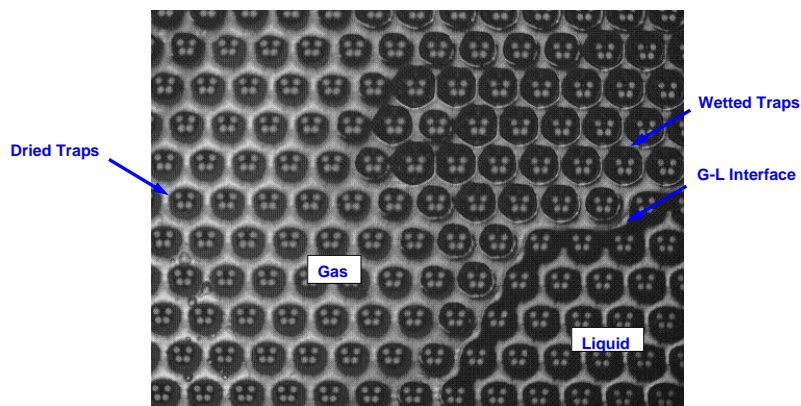
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Catalyst Trap Microreactor

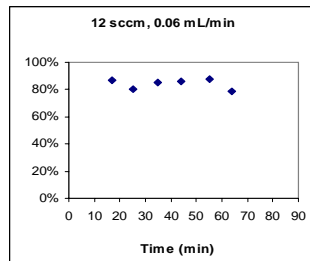


Wetted Traps



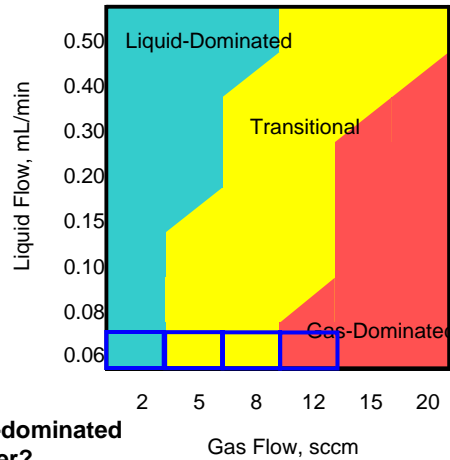
- Indication of the type of flow regime present.
- Traps function as a network of "mini-batch" reactors. Measured conversion exceeds model prediction.

Conversion Results (in progress)



Still limited by G/L area?

Can we achieve higher yields in L-dominated region where productivity is greater?



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Key Limitations in MCS and Examples of How We Are Exploring Them

- Geometry Limitation in Heat Removal
- Thermal Integration Limitation in a Multiunit System
- Multiphase Mixing Limitation
- **Integrated Process Control Limitation**

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Integrated Process Control Needs for Portable MCS-Based Systems

- Integrated sensing (temperature, flow, pressure, etc.)
- Integrated actuation (valves, heaters, etc.)
- Control strategy
- Miniature
- Autonomous

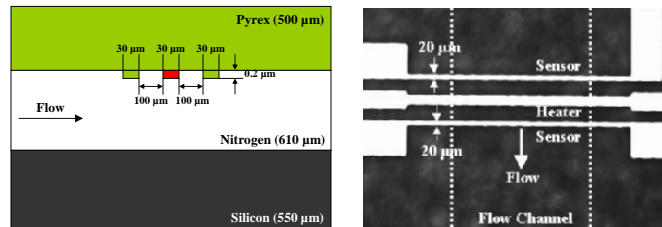
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We Need a MicroFlow Sensor. Build One.

Principle of a Simple Flow Sensor: Calorimetric Sensor

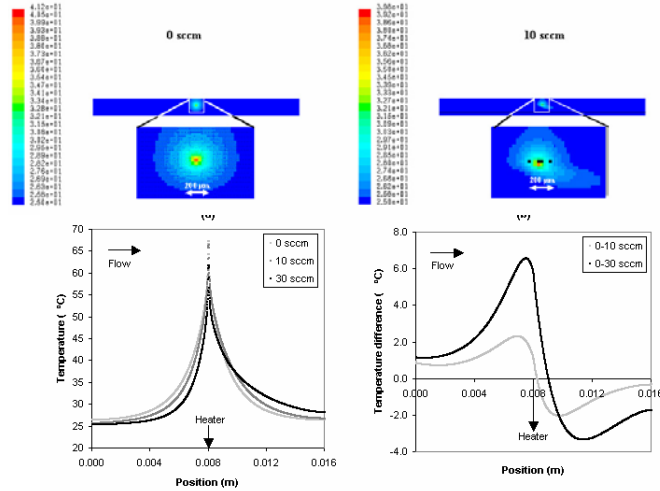
1. Heat generation by Joule dissipation in center resistor.
2. Trigger temperature changes of sensing resistors by convective movement of the fluid stream.
3. Temperature changes lead to resistance changes.
4. Monitoring resistance changes through voltage drop.



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Will It Work and How (CFD Simulation)



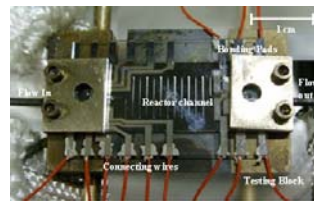
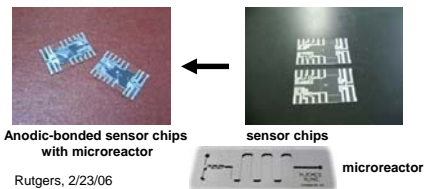
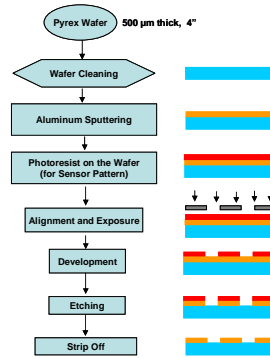
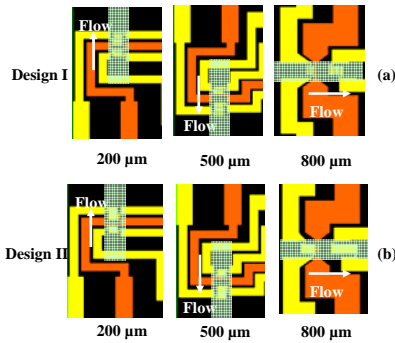
Temperature profiles with different flow rate

Temperature difference between static and flow conditions

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Design & Fabrication



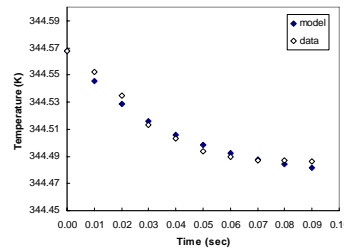
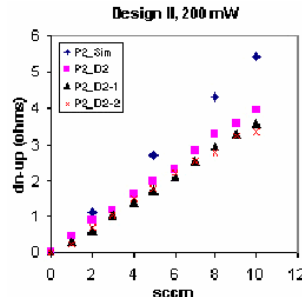
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Flow Sensor Performance Limitations

Sensitivity
 Limited primarily by resistance of sense resistors within thermally affected zone

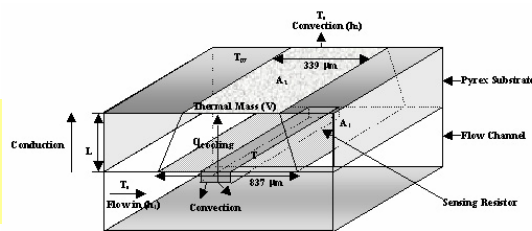
Response time $\cong 70$ msec
 Limited primarily by mass of Pyrex substrate above thermally affected zone



Flow Sensor Development (Transient Thermal Response)

Lumped Thermal Capacitance Model:

Biot number $\ll 1$
 (Internal Conduction Resistance/
 External Convection Resistance)



Thermal model of sensor

Heat loss from thermal mass = Convection to gas flow + Conduction to substrate

$$Vc\rho\left(\frac{dT}{dt}\right) = q_{convection} + q_{conduction}$$

$$Vc\rho\left(\frac{dT}{dt}\right) = -A_1h_1(T - T_s) - LK_{py}(T - T_{py})$$

Flow Sensor Development Summary

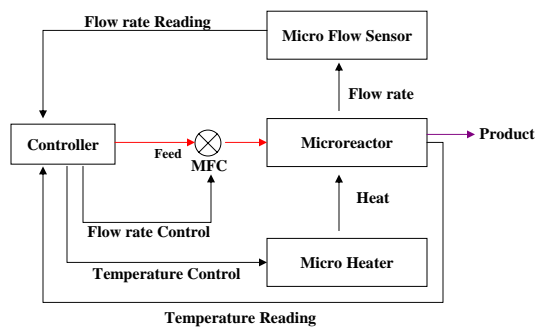
- Satisfactory sensitivity
- Matched to the flow range of microreactor (10 sccm)
- Limitations on sensitivity
 - R of the sensing resistor, separation of heater and sensor, and the heating power
- Limitations on response time
 - With reduction of Pyrex thickness, response time could be reduced to < 10 msec
- Integratability, manufacturability, cost

W.C. Shin and R.S. Besser, "Micromachined Thin-film Gas Flow Sensor for Micro Channel Chemical Reactor," J. of Micromechanics and Microengineering, 2006, in press.

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A Control Scheme is Needed



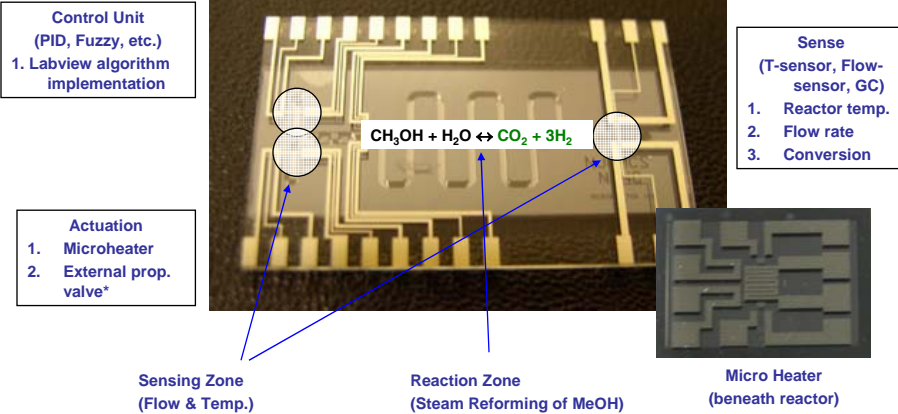
Block diagram of feedback control system for microreactor system

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Microchemical System Control

Experimental Plan



1. Perform the reaction with different control schemes. (Fuzzy logic, PID, etc.)
2. Demonstrating effective control scheme and condition for SR-reaction.

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Geometric Limitation on Heat Transfer

1. 500 μm vs. 2 mm made a big difference

Thermal Integration of Separate Units

1. Minimum design 1.3 W

Multiphase Mixing

1. Do novel flow regimes offer an advantage?

Miniature Integrated Control

1. Limitations of integrated flow sensor manageable.

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Microchemical Systems

1. Surface to volume ratio

Acknowledgements

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Prof. Woo Lee, SIT
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Dr. Art Kaufman
X. Ouyang, Ida-Tech
L. Bednarova, Ultracell

Agencies, Institutions

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Defense Advanced Research
Projects Agency
New Jersey Commission on
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Stevens Institute of Technology
TACOM-ARDEC
Office of Naval Research
Cornell Nanofabrication Facility
(NSF)

People (cont.)

K. Shah, W. Shin
S. McGovern, H. Gadre